17314

21718 3 Hours / 100 Marks

Seat No.

Instructions : (1) All Questions are *compulsory*.

- (2) Illustrate your answers with neat sketches wherever necessary.
- (3) Figures to the right indicate full marks.
- (4) Assume suitable data, if necessary.
- (5) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

$6 \times 2 = 12$ 1. (A) Answer any SIX : Write down any four industrial applications of sulphuric acid. (a) State raw materials used for manufacturing of sulphuric acid. (b) Give reason SO_3 dissolved in conc. H_2SO_4 and not in water. (c) Enlist four properties of sulphuric acid. (d) State merits of contact process. (e) Enlist four industrial uses of ammonia. (f) List two methods of production of cement. (g) Write the cell notation for diaphragm cell. (h)

[1 of 4] P.T.O.

(B) Answer any TWO :

- (a) Write the names of raw material required to manufacture HCl by synthesis process and give its manufacturing flow diagram.
- (b) State Linde's and Claude's principle for manufacturing of oxygen and nitrogen.
- (c) Describe manufacturing of Portland cement by dry process with a neat process flow diagram.

2. Answer any TWO :

- (a) Explain manufacturing process of nitric acid with its flow diagram.
- (b) Describe with flow sheet process for manufacture of triple super-phosphate.
- (c) Draw and describe manufacturing process of chlorine and caustic soda by electrolytic process.

3. Answer any FOUR :

- (a) Distinguish between yellow and red phosphorus.
- (b) Explain the manufacturing of H_3PO_4 by wet process. Draw diagram with description.
- (c) Draw flow sheet for phosphorus trichloride manufacturing. Write reactions involved.
- (d) Draw a well labelled diagram of mercury cell.
- (e) Write two properties and two uses each of caustic soda and chlorine.
- (f) Give any four industrial uses of soda ash.

17314

 $4 \times 4 = 16$

 $2 \times 8 = 16$

17314

4. Answer any FOUR :

- (a) Name raw material, reaction used in manufacturing of soda ash.
- (b) Explain how yellow phosphorus is converted to red phosphorus.
- (c) Compare between dry and wet process for cement manufacture. (any 4 points)
- (d) Draw a furnace used in production of phosphorus.
- (e) Describe manufacturing process of water gas.
- (f) Enlist 4 uses of hydrogen.

5. Answer any TWO :

- (a) Describe with flow sheet Stengel process for ammonium nitrate.
- (b) Describe manufacturing process for acetylene from calcium carbide. Draw the flow diagram for the process.
- (c) Name two mixed fertilizers. What do you mean by N.P.K. ? Why mixed fertilizers are popular now-a-days ?

6. Answer any FOUR :

- (a) Draw block diagram of manufacture of carbon dioxide.
- (b) What is plaster of paris ? Give its uses.
- (c) What is biurate ? State any two uses of urea.
- (d) Draw a neat flow diagram for manufacturing of producer gas.
- (e) Compare between triple and single superphosphate on the basis of raw materials, uses, reaction and process.
- (f) Describe manufacturing of ammonium phosphate.

$2 \times 8 = 16$

 $4 \times 4 = 16$

17314