

Subject Code:

Important Instructions to examiners:

- 1) The answers should be examined by key words and not as word-to-word as given in the model answer scheme.
- 2) The model answer and the answer written by candidate may vary but the examiner may try to assess the understanding level of the candidate.
- 3) The language errors such as grammatical, spelling errors should not be given more Importance (Not applicable for subject English and Communication Skills.
- 4) While assessing figures, examiner may give credit for principal components indicated in the figure. The figures drawn by candidate and model answer may vary. The examiner may give credit for any equivalent figure drawn.
- 5) Credits may be given step wise for numerical problems. In some cases, the assumed constant values may vary and there may be some difference in the candidate's answers and model answer.
- 6) In case of some questions credit may be given by judgement on part of examiner of relevant answer based on candidate's understanding.
- 7) For programming language papers, credit may be given to any other program based on equivalent concept.

Q. No.	Sub Q. No.	Answer			
1	a)	Attempt any six of the following	12		
1a)	i)	What is isentropic process? Plot it on P-V diagram.			
	Answer:	Isentropic Process: The reversible adiabatic or frictionless adiabatic process is known as Isentropic Process. It is a constant entropy process. P 1 1 P V 1 P V Diagram	1		
		Isentropic process			
1a)	ii)	State why multistaging is necessary in air-compressor			
	Answer:	Necessity of multistaging with intercooling in air compression: The large pressure ratio gives rise in high compression ratio and high discharged temperature which produce adverse effect on the efficiency and performance of the system. In such application efficiency decreases and works done and power increases. So to get better performance and saving in work and power multistaging with intercooling is necessary.	2		



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1a)	iii)	List out merits of liquid fuels over gaseous fuels.				
	Answer:	 i) Comparatively Less space required to storage. ii) Comparatively Less chance of explosion. iii) There is no loss of heat during storage. iv) Comparatively less inflammable. v) Comparatively easy to handle. vi) Comparatively low cost. 	For any four ^{1/2} mark each			
1a)	iv)	Define Piston Displacement related to air compressor.				
	Answer:	Piston Displacement: It is the volume swept through by the piston in cubic meter per minute for single acting compressor.For double acting compressor it is the volume swept through by both the sides of piston.				
1a	v)	Define Wet steam and Superheated steam.				
	Answer:	 Wet steam- When steam contains some water particles in suspended form, is known as wet steam. The suspended water particles are known as moisture . Superheated steam: If dry saturated steam is heated beyond its saturation temperature, its temperature will increase and this steam is called as superheated steam. Superheated steam have no moisture content 	1			
1a)	vi)	List out Applications of gas turbine.				
	Answer:	Applications of gas turbine: (Any four) 1. Supercharging of I.C. engine 2. For locomotive Propulsion 3. Ship Propulsion 4. Industrial application 5. Air craft engine 6. Electric power generation 7. Turbo-jet engine 8. Turbo-prop engine 9. Ram-jet engine 10. Pulse-jet engine	For any four ^{1/2} mark each			
1a	vii)	State disadvantages of conventional energy sources				
	Answer:	 disadvantages of conventional energy sources 1. Once a conventional energy source is used up it cannot be replaced again. 2. conventional energy source are highly polluting sources and increase the greenhouse gasses. 3. The conventional energy source are responsible for all kinds of non-biodegradable material accumulation. 4. The exposure to conventional energy sources has increased the level of pollution. 5. The rise in temperature due to greenhouse gas accumulation. 	1 Mark each For any two points			



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1a)	viii)	What is Calorific value" of fuel? Define High Calorific value.	2
	Answer:	"Calorific value" of fuel:It is defined as the amount of heat liberated during complete combustion of 1 kg of fuel.It is expressed in terms of KJ/kg.	1 Mark
		H.C.V. of Fuel: Higher calorific value of fuel is defined as amount of heat energy obtain by the complete combustion of 1kg of fuel, when the products of its combustion are cooled down to the temperature of supplied air.	1Mark
1	b	Attempt any two of the following	8
1b)	i)	Represent isobaric and isochoric process on P-v and T-S diagram	4
		P = C (a) $P = V$ (a) $P = V$ (a) $T = S$ (b)	2
		2) Isochoric Process: $\begin{array}{c} 1 \\ 1 \\ 1 \\ 1 \\ \end{array}$ P-V Diagram T-S Diagram	2
1b)	ii)	Describe the different phases of formation of steam.	4
	Answer:	(Note: Description:2 marks and Diagram:2 marks.)	
		Different phases of Formation of steam- Consider formation of steam from ice at -100 C	
		i) Solid phase- When the heat is added in ice which is at -100 C, the temperature of ice increases to 00 C as shown in figure by process a-b.in this stage solid phase exists.	2



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		Processes in Brayton Cycle:		
		1. Process 1-2: Reversible Adiabatic Compression process		
		2. Process 2-3: Heat addition at constant Pressure		
		3. Process 3-4: Reversible Adjabatic Expansion process	2	
		4 Process 4-1: Heat rejection at constant Pressure	4	
2	с	Define the following	4	
		i) Dryness fraction ii) Degree of superheat		
		iii) Sensible heat iv) Latent Heat		
	Answer:	i) Dryness Fraction: It is defined as a fraction of steam that is in vapour form	1	
		in liquid vapour is called dryness fraction.		
		Mv		
		x =		
		Mv + ML		
		Where		
		x – Dryness fraction		
Mv – mass of vapour (dry steam) contain in steam				
	ML = mass of water in suspension in steam			
		Dryness fraction is ratio of the mass of actual dry steam to the mass of wet steam.	1	
		ii) Degree of Superheat: It is difference between the temperature of	-	
		Superheated vapour and the saturation temperature correspondingly to		
		given pressure is said to be Degree of Superheat.		
		iii) Sensible heat: It is defined as the quantity of heat which can be sensed by the	1	
		thermometer. OR The amount of heat added up to saturation temperature is	1	
		called sensible heat.		
		iv) Latent heat: It is defined as the quantity of heat required for phase change of		
		working substance at saturation temperature. OR The amount of heat added	1	
		at saturation temperature is called latent heat.	1	
2	d	Calculate the enthalpy of 1 kg of steam at a pressure of 8 bar and dryness fraction of	4	
	-	0.8. How much heat would be required to raise 2 kg of this steam from water at	-	
		20° C? Assume P=8bar, hf= 720.9 kJ/kg, h _{fg} = 2046.5 kJ/kg		
	Answer:	Given :-		
		1) For 1 Kg of steam		
		m=1Kg		
		x=0.8		
		h _f =720.9 kJ/kg		
		$h_{fo} = 2046.5 \text{ kJ/kg}$		
		$\dot{\mathbf{h}} = \mathbf{m}(\mathbf{h}_{f} + \mathbf{x} \mathbf{h}_{f\sigma})$	1	
		$h=1[720.9 + (0.8 \times 2046.5)]$	T	
		Enthalpy		
		h = 2358.1 KJ	1	
			T	



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		 2) Heat required for 2Kg of steam from 20°C of water m= 2Kg C_p=4.2 KJ/KgK This value is of enthalpy at 0°C. For 20°C heat in the water= specific heat of water X Rise in temperature = 4.2 X 20 = 84 KJ Heat required per Kg of steam= 2358.1 – 84 = 2274.1 KJ 			
		And			
2		Heat required of 2 Kg of steam= 2 X 2274.1 = 4548.2KJ Explain different modes of heat transfor	1		
	e	Explain different modes of heat transfer	4		
	Answer :	 Conduction- It is the mode of heat transfer from one part of substance to another part of same substance or one substance to another without displacement of molecules or due to the vibrations of molecules. Example-Heat transfer in metal rod. Convection: It is the mode of heat transfer from one part of substance to another part of same substance or one substance to another with displacement of molecules or due to the fluid flowing. Example: Heat flow from boiler shell to water. Radiation: It is the transfer of heat through space or matter. For Radiation there is no need of medium as like convection and conduction. It passes through vacuum in the form of all the passes through vacuum in the	4		
		Example: The energy from sun to the earth surface.			
2	f	Represent Otto cycle and Diesel Cycle on P-V and T-s Diagram and Write equation for Air Standard Efficiency of Otto cycle.	4		
		1. Otto cycle $ \int_{(a)} u_{rest} \int_{(b)} u_{rest} \int_{(c)} u_{rest} \int_{(c$	1 ¹ /2 Mark		



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3	f)	A coal has the following composition by mass: C= 85%, H2= 4%, S= 1% O2 =2%, N2 =1%, and remaining is Ash. Find HCV and LCV of fuel.		
		Given Data: Carbon C = $85\% = 0.85$ Hydrogen = H2 = $4\% = 0.04$ Oxygen = O2 = $2\% = 0.02$ Nitrogen = N = $1\% = 0.01$ Sulphur = S = $1\% = 0.01$ Ash = $7\% = 0.07$ Dulong's formula:	02	
		H.C.V. of coal = $33800 \text{ C} + 144500 (\text{H}_2 - \text{O}_2/8) + 9300 \text{ S} \text{ KJ} / \text{Kg}$ = $33800 \text{ x} 0.85 + 144500 (0.04 - 0.02/8) + 9300 \text{ x} 0.01$ = $34241.75 \text{ KJ} / \text{Kg}$		
		L.C.V. of coal = H.C.V $9H_2 \times 2442 \text{ KJ} / \text{Kg}$ = $34241.75 - (9 \times 0.04) \times 2442$ = $33362.63 \text{ KJ} / \text{Kg}$	02	
04		Attempt any TWO of the Following	16	
4	a)	Attempt the following	8	
4a)	i)	Explain working of Geothermal Power plant with the help of neat sketch.		
	Answer:	i) Working of geothermal power plant		
		Steam drum atmosphere Generator Turbine Steam Steam Turbine Ground level to ground	02	
		Figure shows geothermal power plant which consists of the following main components: Underground steam storage, steam separator, steam separator, turbine and Generator. Steam is present in the earth crust at 10 km depth is about 2000 C. It is stored in the underground steam storage tank. This steam is taken out through pipe and valve and passed through steam separator. In steam separator moisture content in the steam is taken out and dry steam is allowed to pass in steam	02	



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		drum where steam is stored. The moisture content in steam is then injected into the ground. As per requirement steam is passed over the turbine and kinetic energy of steam is converted into mechanical work. Turbine is connected to the generator by shaft which generates power. Mechanical energy of shaft is converted in to electrical energy by generator.					
4a)	ii)	Explain construction and working of Bomb calorimeter.	4				
	Answer:	Construction : The calorimeter is made of austenitic steel which provides considerable resistant to corrosion and enables it to withstand high pressure. In the calorimeter use of a strong cylindrical bomb in which combustion occurs. The bomb has two values at the top. One supplies oxygen to the bomb and other releases the exhaust gases. A crucible in which a weighed quantity of fuel sample is burnt is arranged between the two electrodes as shown in fig. The calorimeter is fitted with water jacket which surrounds the bomb To reduce the losses due to radiation calorimeter is further provided with a jacket of water and air. A stirrer for keeping the temperature of water uniform and a thermometer the temperature up to accuracy of 0.0010 C is fitted through the lid of the calorimeter.	1				
		Working : The calorimeter is made of austenitic steel which provides considerable resistant to corrosion and enables it to withstand high pressure. In the calorimeter use of a strong cylindrical bomb in which combustion occurs. The bomb has two values at the top. One supplies oxygen to the bomb and other releases the exhaust gases. A crucible in which a weighed quantity of fuel sample is burnt is arranged between the two electrodes as shown in fig. The calorimeter is fitted with water jacket which surrounds the bomb To reduce the losses due to radiation calorimeter is further provided with a jacket of water and air. A stirrer for keeping the temperature of water uniform and a thermometer the temperature up to accuracy of 0.0010 C is fitted through the lid of the calorimeter. The heat released by the fuel on combustion is absorbed by the surrounding water and the calorimeter. From the above data the calorific value of the fuel can be found.	1				



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		$\frac{T_2}{T_1} = \frac{P_2 V_2}{P_1 V_1} \dots \dots$	02
		From (1)	
		$\frac{V_2}{V_1} = \left(\frac{P_1}{P_2}\right)^{1/\gamma}(3)$	
		Put equation (3) into equation (2)	
		$\frac{T_2}{T_1} = \frac{P_2}{P_1} \left(\frac{P_1}{P_2}\right)^{1/\gamma}$	
		$\frac{T_2}{T_1} = \left(\frac{P_2}{P_1}\right)^{\frac{\gamma-1}{\gamma}}$	
		$\frac{P_2}{P_1} = \left(\frac{T_2}{T_1}\right)^{\frac{\gamma}{\gamma-1}} \dots $	02
		From equation (1) & (4)	
		$\frac{P_2}{P_1} = (\frac{V_1}{V_2})^{\gamma} = (\frac{T_2}{T_1})^{\frac{\gamma}{\gamma-1}}$	
		$\frac{P_2}{P_1} = (\frac{V_1}{V_2})^{\gamma} = (\frac{T_2}{T_1})^{\frac{\gamma}{\gamma-1}}$	02
4	C)	Evaluin the construction and working of	8
4c)	i)	Explain the construction and working of:	4
		Centrifugal compressor.	
	Answer:	Centrifugal compressor:	
		Fig. shows centrifugal compressor, it is simple in construction. It consists	
		of rotor (i.e. impeller), impeller eye and diffuser. In impeller number of curved	
		vanes is fitted symmetrically. Impeller rotates in an air tight volute casing. The	
		casing is designed that the kinetic energy of the air is converted into pressure	
		energy before it leaves the casing. Mechanical energy is provided to impeller by	
		some external means. As impeller rotates it sucks air from impeller eye, increases	2
		Its pressure due to centrifugal force and forces the air to flow over diffuser. The	
		high pressure of air further increases during its flow over diffuser. Finally, the air at high pressure is delivering to receiver. The air enters in the impeller radially and leaves vanes axially.	



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		Delivery	
		Casing Diffuser Direction of rotation Impeller eye	2
4c)	ii)	Axial flow Compressor:-	4
		Axial flow compressor: The basic components of an axial flow compressor are a rotor and stator, the former carrying the moving blades and the latter the stationary rows of blades. The stationary blades convert the kinetic energy of the fluid into pressure energy, and also redirect the flow into an angle suitable for entry to the next row of moving blades. Each stage will consist of one rotor row followed by a stator row, but it is usual to provide a row of so called inlet guide vanes. This is an additional stator row upstream of the first stage in the compressor and serves to direct the axially approaching flow correctly into the first row of rotating blades. For a compressor, a row of rotor blades followed by a row of stator blades is called a stage. In an axial compressor, the flow rate tends to be high and pressure rise per stage is low. It also maintains fairly high efficiency.	2
		(Variable) stator vanes	2



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Model Answer

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		3 Temperature of condensate is more than of food water so amount of heat				
		5.	supplied per kg of steam is reduced	te than of feed water so amount of heat		
		Location : It locates in between Turbine and Feed pump				
5	b)	Differentiate between reciprocating and rotary air compressor.			08	
	Answer	Differe	ence between reciprocating and ro	tary air compressor (any 8 points)	1	
	:	Sr.	Reciprocating air compressor	Rotary air compressor	mark	
		No.			for	
		1	It is having to and fro motion.	It is having rotary motion	each	
		2	Air supply is intermittent.	Air supply is continuous.	point	
		3	Lubrication system is complicated.	Lubrication system is simple.		
		4	Maximum delivery pressure is upto 1000 bar.	Maximum delivery pressure is upto 10	(Any 8 Points	
		5	Maximum free discharge is about 300 m ³ /min.	Maximum free discharge is about 3000 m ³ /min)	
		6	Speed is lesser	Speed is higher		
		7	Balancing is major problem	No balancing problem		
		8	Frictional losses are more	Frictional losses are less		
		9	Size of compressor is large for the	Size of compressor is small for the give		
			given discharge	discharge		
		10	It is suitable for low discharge and	It is suitable for high discharge and low		
			high pressure	pressure		
		11	Application- Auto workshop, servio	ce Application- Torbocharger, supercharge		
			stations, air brake system etc.	Blower, Hair Drier etc.		
5	c)	During a boiler trial coal analysis on mass basis was reported as C=62.4%, H_2=4.2%, O_2=4.5%0Moisture=15% and ash=13.9%. Calculate minimum air required to burn 1 kg of coal. Also calculate Higher and lower calorific value.0				
		Given-	C=62.4%=0.624 Kg H ₂ =4.2%=0.042 Kg O ₂ =4.5%=0.045 Kg Moisture=15%=0.15 Kg Ash=13.9%=0.139 Kg			
		Minin	uum air required to burn 1 kg of coal	$=\frac{100}{23}\left[\left(\frac{8}{3}C+8H_2+S\right)-O_2\right]kg$ $100\left[\left(\frac{8}{3}C+8H_2+S\right)-O_2\right]kg$	3	
				$= \frac{1}{23} \left[\left(\frac{1}{3} \times 0.024 + 8 \times 0.042 \right) - 0.045 \right]^{\frac{1}{3}}$		
				= 8.67 Kg		



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		H.C.V. of coal = $33800 \text{ C} + 144500 (\text{H}_2 - \text{O}_2/8) + 9300 \text{ S} \text{ KJ / Kg}$ = $33800 \times 0.624 + 144500 (0.042 - 0.045/8)$ = 26348.11 KJ / Kg						
		L.C.V. of coal = H.C.V $9H_2 \times 2442 \text{ KJ} / \text{Kg}$						
		= 26348.11 - 9X 0.042X 2442						
		= 25425.034 KJ / Kg	2					
6		Attempt any <u>FOUR</u> of the following	16					
6	a)	What are applications of heat transfer in automobile?	4					
		Answer:Applications (any four) i)Radiator ii)Engine with components as cylinder liner, piston and piston rings, cylinder heads, exhaust valves etc. iv)Lamps v)Cooling circuits vi)Exhaust system vii)Lubricating system	1 Mark each any four					
6	b)	Explain the sources of air leakages in condenser.	4					
		 Answer: Sources of air leakage in condenser 1. Air leak through joints and packing. Air leaks into condenser as pressure inside falls below atmospheric pressure. 2. Air also comes in condenser with the steam. The feed water supplied to the boiler contains certain amount of air dissolved in it. The dissolved air gets liberated when steam is formed and is carried with the steam into the condenser. 3. In jet condensers dissolved air in the cooling water enters the condenser. The dissolved air gets separated at low pressure in the condenser 4. Air leaks if any bypass seal is broken. 	1 Mark each any four					



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		6 7 8 9 10	The exhaust gas from the turbine is passed into cooling chamber This system required more space. Heat exchanger is used. Compressed air is heated in air heater It is suited for stationary installation, marine use.	The exhaust gas from the turbine is exhausted to the atmosphere.This system required less spaceHeat exchanger is not usedCompressed air is heated in combustion chamberIt is suited for moving vehicle.	
6	e)	Discus	s solar energy as non-conventional en	ergy source.	4
		(<i>Note:</i> From t current energy conven availab source.	Credit should be given to equivalent and Solar energy is very large, inexhaust he sun we receive about 10^{14} Kw of energy used from all sources. So solar energy needs of the world on a continuing basis tional source it is an environmentally cl le in adequate in almost all parts of the This makes it is one of the most promise	<i>iswer)</i> ible, readily available source of energy. rgy which is 5 times greater than could supply all the present and future s. In addition to this unlike ean source of energy and it is free and world. sing of the non-conventional energy	4
6	f)	Discus	s CNG and LPG as gaseous fuels.		4
		CNG-It is called as compressed natural gas. For transportation, storage and automotive use natural gas which is mainly composed of methane(CH4) is compressed and stored in high pressure cylinder. CNG is used in traditional gasoline internal combustion engine cars that have been converted into bi-fuel vehicles(gasoline/CNG). It's Calorific Value is 46000 to 49000 KJ/kg. LPG- It is called as liquefied Petroleum Gas. It consist of butane, propane and exist in gaseous form under atmospheric condition but can be easily liquefied under pressure. It is used in domestic purpose as well as rural heating, Motor vehicle. It's Calorific Value is 46100 KJ/kg.2			