

17410

16172

3 Hours / 100 Marks

Seat No.

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- Instructions* –
- (1) All Questions are *Compulsory*.
 - (2) Answer each next main Question on a new page.
 - (3) Illustrate your answers with neat sketches wherever necessary.
 - (4) Figures to the right indicate full marks.
 - (5) Assume suitable data, if necessary.
 - (6) Use of Non-programmable Electronic Pocket Calculator is permissible.
 - (7) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.
 - (8) Use of Steam tables, logarithmic, Mollier's chart is permitted.

Marks

1. a) **Attempt any SIX of the following:** **12**
- (i) Define path function and point function.
 - (ii) State Clausius statement of second law of thermodynamics.
 - (iii) Represent isobaric process on P-V and T-S charts.
 - (iv) State the relationship between universal gas constant and characteristic gas constant. Write the meaning of each term.
 - (v) Differentiate between boiler mountings and accessories (any two points)

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- (vi) State the meaning of the terms 'governing' and 'compounding' of steam turbines.
- (vii) Write continuity equation of steam nozzle and state the meaning of each term.
- (viii) Define condenser efficiency.
- b) **Attempt any TWO of the following:** **8**
- (i) Define following terms:
- 1) Dryness fraction
 - 2) Degree of superheat
 - 3) Dry saturated steam
 - 4) Superheated steam
- (ii) State Dalton's law of partial pressures. Apply it to steam condenser with the help of suitable diagram.
- (iii) Define black body, grey body, emissivity and absorptivity.
2. **Attempt any FOUR of the following:** **16**
- a) Differentiate between heat engine and heat pump. (any four points)
- b) A tank of 2.2 m^3 capacity contains air at 260°C and 0.1 MPa . Some air is taken from the tank without changing temperature until pressure becomes 4 KPa . Calculate:
- (i) Mass of air left in the tank
 - (ii) Mass of air pumped out. (Take $R = 0.287 \text{ KJ/kg}^\circ\text{K}$)
- c) What is boiler draught? State various types of boiler draughts with meaning.
- d) Compare impulse and reaction turbine on the basis of following points:
- (i) Shape of blade
 - (ii) Admission of steam
 - (iii) Power generated
 - (iv) Speed
- e) Enlist various losses in steam turbines.
- f) Write steady flow energy equation and apply it to boiler and nozzle.

3. Attempt any FOUR of the following: 16

- a) State and explain various types of thermodynamic systems with examples.
- b) Represent isothermal and adiabatic processes on P-V and T-S diagrams.
- c) Draw neat sketch of Cochran boiler. Label all the parts.
- d) Explain nozzle control governing of steam turbine with neat sketch.
- e) Compare jet and surface condensers on the basis of following points:
 - (i) Amount of cooling water required
 - (ii) Vacuum efficiency
 - (iii) Construction
 - (iv) Operation of heat exchange
- f) What is heat exchanger? Give three examples of heat exchangers with their applications in thermal engineering.

4. Attempt any FOUR of the following: 16

- a) Prove the equivalence of Kelvin Planck and Clausius statements of second law of thermodynamics.
- b) Represent steam generation process at constant pressure on T-S diagram and show following on it:
 - (i) Saturated liquid line
 - (ii) Saturated vapour line
 - (iii) Critical point
 - (iv) Wet Region
- c) Explain working principle of reaction turbine with suitable sketch.
- d) A composite wall is made up of 0.2 m thick fire clay brick, faced with 0.1 m thick insulation. If temperatures of inner and outer surfaces are 870°C and 210°C respectively, calculate heat flow rate per unit area. Also find interface temperature.
Take $K_{\text{brick}} = 1.039 \text{ W/m}^\circ\text{k}$, $K_{\text{insulation}} = 0.228 \text{ W/m}^\circ\text{k}$.

- e) Vacuum measured at the inlet of condenser is 710 mm of Hg and barometer reads 760 mm. Hot well temperature is 30°C. Calculate vacuum efficiency.
- f) Write the formulae to calculate enthalpy and entropy of the following:
 - (i) Wet steam
 - (ii) Superheated steam

5. Attempt any TWO of the following: 16

- a)
 - (i) Differentiate between heat and work.
 - (ii) Explain zeroth law of thermodynamics.
- b) Sketch and explain pressure compounded and velocity compounded impulse turbine showing pressure and velocity variations along the axis.
- c) A certain gas has $C_p = 1.968 \text{ KJ/kg}^\circ\text{K}$ and $C_v = 1.507 \text{ KJ/kg}^\circ\text{K}$. Find its molecular weight and gas constant. A constant volume chamber of 0.3 m^3 capacity contains 2 kg of this gas at 5°C. Heat is transferred to gas until temperature is 100°C. Find work done, heat transfer and change in internal energy.

6. Attempt any TWO of the following: 16

- a)
 - (i) Draw neat sketch of any one type of surface condenser. Label parts.
 - (ii) State any two sources and effects of air leakage into steam condenser.
 - b)
 - (i) A sample of 3 kg of steam at a pressure of 3 MPa exists in dry and saturated condition. For this sample, calculate enthalpy and entropy using steam table.
 - (ii) Steam at 8 bar and 0.85 dry is throttled to a pressure of 1 bar. Find final quality of steam. Use steam table.
 - c) Explain various modes of heat transfer with suitable examples.
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